

Excited-State Structure Modifications Due to Molecular Substituents and Exciton Scattering in Conjugated Molecules

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ABSTRACT: Attachment of chemical substituents (such as polar moieties) constitutes an efficient and convenient way to modify physical and chemical properties of conjugated polymers and oligomers. Associated modifications in the molecular electronic states can be comprehensively described by examining scattering of excitons in the polymer's backbone at the scattering center representing the chemical substituent. Here, we implement effective tight-binding models as a tool to examine the analytical properties of the exciton scattering matrices in semi-infinite polymer chains with substitutions. We demonstrate that chemical interactions between the substitution and attached polymer are adequately described by the analytical properties of the scattering matrices. In particular, resonant and bound electronic excitations are expressed via the positions of zeros and poles of the scattering amplitude, analytically continued to complex values of exciton quasimomenta. We exemplify the formulated concepts by analyzing excited states in conjugated phenylacetylenes substituted by perylene.

SECTION: Spectroscopy, Photochemistry, and Excited States

The concept of scattering has proven to play an important I role in the description of a variety of phenomena in quantum mechanics, condensed matter and quantum field theory, functional analysis, and chemistry. Scattering matrices/ operators contain detailed information on fundamental interactions that control the system dynamics and are available from experimental measurements. In high-energy physics, the differential cross sections, directly related to scattering matrices, are measured and further interpreted using quantum field theory, with the ultimate goal of revealing the fundamental interactions. In condensed matter theory, the scattering matrices provide a useful tool of resumming the short-range interaction effects, for example, in the superfluidity theory in the low-density case, where a perturbation theory can be formulated in terms of the particle-particle scattering matrix that contains all necessary information on the particle–particle interactions.^{1–3} The celebrated Fermi liquid theory allows interpretation of complex dynamics of a strongly interacting system to be interpreted in terms of quasi-particle spectra and scattering matrices.^{4,5} Chemical reactions in the gas phase can be conveniently formulated and interpreted as scattering processes with the asymptotic states represented by the reactants and products.^{6,7} In functional analysis, scattering theory can be viewed as a tool of analyzing continuous spectra of unbounded (often differential) operators, in particular, the projection measures involved in spectral decompositions.⁸

Our recent work $^{9-16}$ has demonstrated that the quasi-particle picture, coined exciton scattering (ES) approach, provides



simple and clear insight into the excited-state electronic structure in complex conjugated macromolecules.¹⁷⁻²¹ It allows excited electronic states to be studied in terms of the exciton spectra in infinite polymers and scattering matrices to be associated with molecular vertices, that is, termini, joints, and branching centers.^{22–28} The ES properties of molecular vertices can be further described by tight-binding or equivalent lattice models.²⁹ This extends the ES concept to the case of imperfect molecular geometries aimed at deriving the exciton-phonon Hamiltonian, thus mapping the problem of incoherent energy transfer in branched conjugated structures onto a much simpler (although still complex) counterpart of incoherent motion of Frenkel-type excitons.^{30,31} The scattering framework in the conjugated molecular systems is different from typical cases usually studied in quantum mechanics due to discrete rather than continuous translational symmetry of the asymptotic states. In particular, integer topological invariants, namely, winding numbers/topological charges, can be associated with the scattering centers.³² It is well-established in quantum mechanics that analytical properties of scattering matrices, more specifically their analytic continuations, provide detailed and important information on the underlying potentials, including bound and metastable states. Therefore, analytical properties of ES on molecular vertices are expected to provide

Received: December 17, 2013 Accepted: January 28, 2014 Published: January 28, 2014 adequate information of how electronic properties of supramolecular conjugated structures are affected by local chemical substituents.

In this Letter, we apply tight-binding models as a tool to study analytical properties of scattering matrices, providing insights into the excited-state electronic structure of molecular substituents commonly present in conjugated systems. As an example, Figure 1c displays the reflection phase of perylene



Figure 1. (a) Perylene-terminated PA linear molecule. (b) The tightbinding model describing the ES at the terminus. (c) The exciton reflection phase of the terminus obtained from the ES approach (black solid curve), which is approximated by the tight-binding model shown in (b) (red dashed curve) and by the simplified model (blue dotted– dashed curve), where three sharp resonances have been removed. The reflection phase of the unmodified terminus $\phi_{\rm H}$ is shown for comparison.

attached to a semi-infinite phenylacetylene (PA) chain (see Figure 1a), retrieved from quantum chemistry calculations.¹⁵ Compared to that in an unsubstituted chain ($\phi_{\rm H}$), the plot shows highly nontrivial dependence of the phase on the exciton quasi-momentum k, including four resonant features that are not easy to interpret. Here, we show how these features in the reflection phase can be directly ascribed to resonant and bound excited states brought by the substituent.

We start with introducing the simplest nearest-neighbor hopping lattice model, where the linear segments of a branched conjugated structure are represented by linear chains (graphs) with the same on-site energy $\overline{\Omega}_0$ and the same hopping constant \overline{J}_1 between the nearest neighbors. A molecular vertex is represented by a complete graph, that is, by a set of fully interconnected sites, with arbitrary on-site energies and hopping constants between any pair of sites allowed. The chemical connection between a molecular vertex and the attached linear segment is thus described by introducing the hopping constants between the first lattice site (whose on-site energy is also modified) of the linear segment and any lattice site that belongs to the vertex (e.g., see Figure 1b). Here, we consider the case of chemical substitution on the molecular terminus only (e.g., perylene in Figure 1a); the general case of an arbitrary degree molecular vertex will be analyzed elsewhere.

Because the lattice sites representing a vertex form a basis set for the vertex tight-binding Hamiltonian, without loss of generality, we can assume the latter to be diagonal. Therefore, a molecular terminus is described by a set { $\omega_{\alpha}|\alpha = 1, ..., n$ } of the on-site energies, a vector $J = (J_{\alpha}|\alpha = 1, ..., n)$ of the hopping constants between the vertex sites and the first linear segment site, and the modified on-site energy Ω_1 of the latter, as shown in Figure 1b. Measuring the energy in the unit of \overline{J}_1 and choosing the zero energy level at the middle of the exciton band, without loss of generality, we can set $\overline{\Omega}_0 = 0$ and $\overline{J}_1 = 1$.

Within the described lattice model, the exciton wave function on a semi-infinite chain is given by the sets $\Psi = (\Psi_{\alpha}|\alpha = 1, ..., n)$ and $\psi = (\psi_j|j = 0, 1, ...)$ of its values on the terminus and the chain, respectively. Introducing the multiplicative variable $z = e^{ik}$ that describes the quasi-momentum k, we represent the wave function on the chain as a superposition of incoming and outgoing waves

$$\psi_j = z^{-j} + r(z)z^j$$
 $j = 0, 1, ...$ (1)

with r(z) is the quasi-momentum-dependent reflection coefficient at j = 0. The eigenmode equation adopts a form

$$\begin{split} \psi_{j-1} + \psi_{j+1} &= \omega \psi_j \qquad j = 1, 2, \dots \\ \omega_{\alpha} \Psi_{\alpha} + J_{\alpha} \psi_0 &= \omega \Psi_{\alpha} \qquad \alpha = 1, \dots, n \\ \sum_{\alpha} J_{\alpha} \Psi_{\alpha} + \psi_1 &= (\omega - \Omega_1) \psi_0 \end{split}$$
(2)

Upon substituting eq 1 into eq 2, we can easily solve the system of linear equations and obtain the following expressions for the exciton spectrum

$$\omega(z) = z + z^{-1} \tag{3}$$

and reflection coefficient

$$r(z) = -\frac{z - \Omega_1 - F(\omega)}{z^{-1} - \Omega_1 - F(\omega)} \qquad F(\omega) = \sum_{\alpha} \frac{J_{\alpha}^2}{\omega - \omega_{\alpha}}$$
(4)

both being represented by meromorphic functions of z in the complex plane. This implies that r(z) can be also interpreted as a meromorphic function on the projective space $\mathbb{C}P^1$ or a holomorphic function $r:\mathbb{C}P^1 \to \mathbb{C}P^1$. These two interpretations originate from viewing $\mathbb{C}P^1$ as a compact complex analytical manifold (of complex dimension 1), topologically equivalent to a sphere $\mathbb{C}P^1 \cong S^2$ with the complex structure induced from complex plane \mathbb{C} via stereographic projection $\mathbb{C} \to \mathbb{C}P^1$, that is, $\mathbb{C}P^1$ is obtained from \mathbb{C} by adding the infinite point.

The reflection coefficient satisfies obvious relations

$$r(z^{-1}) = (r(z))^{-1}$$
 $r(z^*) = (r(z))^*$ (5)

that reflect unitarity of quantum mechanics combined with time-reversal symmetry.

A direct inspection of eq 4 shows that the reflection coefficient can be represented in a form $r(z) = zP_{2n+1}(z)/Q_{2n+1}(z)$, with P_{2n+1} and Q_{2n+1} being polynomials of degree (2n + 1), which means that r(z), as a meromorphic function in $\mathbb{C}P^1$, has 2(n+1) zeros and 2(n+1) poles or, equivalently, that the map $r:\mathbb{C}P^1 \to \mathbb{C}P^1$ has degree $n_A(r) = 2(n+1)$, hereafter referred to as the analytical index. Due to the symmetry relations (eq 5), the roots of Q_{2n+1} are inverse to the roots of P_{2n+1} , with the roots of each polynomial being either real or coming in mutually complex conjugated pairs. This implies that the positions of the roots, of say P_{2n+1} , fix r(z) up to a multiplicative factor, the latter being fixed by the condition r(1) = -1 [note that also r(-1) = -1] that follows from eq 4 [note that generally eq 5 also implies $r(1) = \pm 1$ as well as $r(-1) = \pm 1$]. Therefore, r(z) can be represented in a form

$$r(z) = z \prod_{j=1}^{m} \frac{w_j z - 1}{z - w_j} \prod_{j=m+1}^{2n+1} \frac{z - z_j}{z_j z - 1}$$
(6)

where $\{z_j\}$ and $\{w_j\}$ are the zeros of the polynomials P_{2n+1} and Q_{2n+1} , respectively, located inside of the circle |z| = 1, hereafter referred to as the circle.

We now tie the analytic properties described above together with the underlying chemistry in terms of bound states. According to quantum mechanical scattering theory and eq 1, the bound states in a semi-infinite chain correspond to the poles of r(z) located inside of the circle, or equivalently the zeros located outside, so that the energies of the bound states are given by $\omega(w_j)$, which implies that w_j should be real for all j= 1, ..., m. In our earlier work (unpublished result), we introduced the topological index (winding number) $n_{\rm T}(r)$, associated with a vertex. A direct calculation yields

$$n_{\rm T} = \oint_{|z|=1} \frac{{\rm d}z}{2\pi {\rm i}} r^{-1} \frac{{\rm d}r}{{\rm d}z} = 2n + 2 - 2m \tag{7}$$

Introducing the molecular vertex analytical and topological charges $Q_A = (n_A - 2)/2$ and $Q_T = (n_T - 2)/2$, respectively, we conclude that in a finite-length linear molecule with *L* repeat units, represented by *L* sites within the nearest-neighbor lattice model, the number of exciton states is given by

$$N = L + Q_{\rm A}^{(1)} + Q_{\rm A}^{(2)} \tag{8}$$

whereas the number of bound states associated with a vertex *a* is given by $m^{(a)} = Q_A^{(a)} - Q_T^{(a)}$. In a molecule with the lengths of linear segments considerably exceeding the localization lengths of the bound states, we have the number of exciton states inside of the exciton band

$$N_0 = L + Q_{\rm T}^{(1)} + Q_{\rm T}^{(2)}$$
(9)

This is true because the excitons with energies outside of the exciton band are excellently approximated by just the bound states. Note that eq 9 reflects one of the statements of the index theorem for the case of linear molecules presented in our earlier work,³³ and eq 7 can be interpreted as a relation between the analytical and topological properties of a molecular vertex; the topological index is formed from (2(n + 1) - m) positive and *m* negative contributions associated with the zeros and the poles, respectively, of r(z), located inside of the circle.

The analytical index theorem (eq 8) can be also derived in more general terms by noting that the ES equations for a linear molecule can be written in a form

$$\tilde{\Gamma}(z) - 1 = 0$$
 $\tilde{\Gamma}(z) = r^{(1)}(z)r^{(2)}(z)z^{2(L-1)}$ (10)

so that the number of its solutions is given by the analytical index $n_A(\tilde{\Gamma} - 1)$ (unpublished result). We further observe

$$n_{\rm A}(\tilde{\Gamma}-1) = n_{\rm A}(\tilde{\Gamma}) = n_{\rm A}^{(1)} + n_{\rm A}^{(2)} + 2(L-1)$$
(11)

and note that there are two unphysical solutions with $z = \pm 1$, whereas each exciton state is represented by a pair of symmetry-related solutions with mutually inverse values of z. This results in $N = [n_A(\tilde{\Gamma} - 1)/2] - 1$, which reproduces eq 8. It is worth mentioning that shifting the reference point of reflection by Δj results in a factor of $z^{2\Delta j}$ in $r(z)^{32}$ and changes of both n_A and n_T by $2\Delta j$, which is followed by the change of Q_A and Q_T by Δj ; the latter is compensated for in eqs 8 and 9 by the corresponding change of L by $-\Delta j$. For convenience, we define Q_A and Q_T using $Q_{A/T} = (n_{A/T} - 2 - 2\Delta j)/2$, so that the analytical and topological charges (integers) are the intrinsic properties of molecular vertices and independent of the reflection point, and the length L should not change with respect to the choice of the reflection position.

In summary, within a nearest-neighbor hopping lattice model, a molecular terminus is described by the reflection coefficient r(z), characterized by its topological charge $Q_{\rm T}$ and analytical charge $Q_A \ge Q_T$, and represented by a meromorphic function on $\mathbb{C}P^1$ in a form given by eq 6. It is fully determined by the positions of its $(Q_A - Q_T)$ poles inside of the circle that all lie on the real axis and $(Q_A + Q_T + 2)$ zeros, referred to as resonances, inside of the circle, which either belong to the real axis or come in mutually complex conjugated pairs. Stated equivalently, using a term "tight-binding model with nearestneighbor hopping" is equivalent to approximating the scattering coefficient r(z) with a meromorphic function, defined in $\mathbb{C}P^1$, because there is a one-to-one correspondence between such meromorphic functions (with the certain simple properties, implied by fundamental quantum mechanical symmetries, and explicitly described earlier in the text) and the sets of parameters of tight-binding models of a certain class, namely, referred to as nearest-neighbor hopping. Therefore, hereafter, we will not make a distinction between the terms "tight-binding model with nearest-neighbor hopping" and "scattering coefficients represented by meromorphic functions in $\mathbb{C}P^{1,n}$. Having said that, we would like to note that dealing with a tight-binding model as a tool has certain advantages by providing simple and intuitive physical insight, as well as opening the way to efficiently account for exciton-phonon interactions, as proposed in our earlier work.²⁹

An interesting situation occurs when a pair (z_{j}, z_{j}^{*}) of resonances lies close to the circle. Introducing a natural notation $z_{j} = (1 - \delta_{j})e^{ik_{j}}$ with $\delta_{j} \ll 1$, we can represent the contribution of the above pair of resonances to the reflection coefficient (eq 4) in a form

$$r_{j}(z) \approx \frac{z - (1 - \delta_{j})e^{ik_{j}}}{z - (1 + \delta_{j})e^{ik_{j}}} \frac{z - (1 - \delta_{j})e^{-ik_{j}}}{z - (1 + \delta_{j})e^{-ik_{j}}}$$
(12)

where we have applied the approximations $(1 - \delta_j)^{-1} \approx 1 + \delta_j$ and $(1 - \delta_j)^2 \approx 1$ (which is irrelevant to the phase factor; therefore, that can be ignored). Except for the narrow regions in the quasi-momentum space of width $\Delta k_j \approx \delta_j$, around $k = \pm k_j$, we have $r_j \approx 1$, that is, it does not contribute to the reflection coefficient, whereas in the above regions, r(z) shows resonant behavior, where the scattering phase $\phi = -i \ln r$ acquires a contribution of 2π over a narrow region $\Delta k_j \approx \delta_j$. In a simplified scenario where $\Omega_1 = 0$, n = 1, and m = 0, one can find that $\delta_1 \approx J_1^2/2$, which indicates that the sharp resonant feature is attributed to the weak coupling between the terminal site and the chain. These resonances, hereafter referred to as phase kinks, correspond to the resonant states, that is, excited states on the substituent, weakly coupled to the exciton band in the polymer chain.

In the aforementioned tight-binding model, the m lattice sites, representing the graph of the terminus, related to the bound states simultaneously result in m poles of r(z) outside of

Table 1.	Tight-Binding	Parameters	(in eV) of	Perylene	Terminus	in PA	Molecules,	Including th	e On-Site	Energy ω_{α}	and
Hopping	g Constant J_{α}										

α	1	2	3	4	5	6	7
ω_{α}	2.716	3.448	3.551	3.713	3.951	4.063	4.209
J_{α}	-0.281	-0.057	-0.047	-0.238	-0.152	-0.032	-0.029

the circle, that is, *m* zeros located inside of the circle. Taking into account a "trivial" zero of r(z) at z = 0 and a zero on the real axis close to z = 0, which is attributed to the small value of Ω_1 , there are 2(n - m) zeros close to and inside of the circle that come in mutually complex conjugated pairs that are associated with the (n - m) resonant states within the exciton band. Stated differently, the $(Q_A - Q_T)$ bound states are directly associated with the poles inside of the circle on the real axis, whereas the resonant states correspond to the Q_T pairs of complex conjugated zeros of r(z) located close to the circle.

To illustrate the above formalism, we calculate electronic excitations in linear PA molecules with and without the perylene substituent (Figure 1a) using a standard quantum chemical (QC) methodology. The length of the PA molecules varies from 5 to 35 repeat units with increments of 5 repeat units. As we have shown before,^{10,16} any approach for excitedstate computation that can adequately describe exciton properties (including the binding energy and the exciton size) can be used as a reference QC method in the ES approach. Here, the ground-state geometries have been optimized at the semiempirical Austin Model 1 (AM1) level³³ using the Gaussian09 package.³⁴ We then applied the collective electronic oscillator (CEO) method, 35-37 which is based on the time-dependent Hartree-Fock (TDHF) theory combined with the semiempirical INDO/S (intermediate neglect of differential overlap parametrized for spectroscopy) Hamiltonian,³⁸ to compute the excitation energies, transition dipoles, and transition density matrices. The lowest exciton band has been singled out by inspecting the structures of transition density matrices in real space.¹² The exciton spectrum $k(\omega)$ and the reflection phases of unmodified and pervlene-substituted termini have been extracted within the ES approach.¹⁵ As shown in our previous studies, physically similar results can be expected using other model QC techniques for excited-state calculations such as time-dependent density functional theory (TDDFT).¹⁶ The ES phase of perylene and the excited-state data provide sufficient input for a corresponding tight-binding model;²⁹ the nearest-neighbor tight-binding parameters in an infinite linear chain, $\overline{\Omega}_0$ and \overline{J}_1 , can be found from the exciton spectrum $k(\omega)$ and eq 3.

We are now in a position to describe the excited-state chemical properties of perylene attached to a PA chain in terms of analytical properties of the reflection phase obtained with a chosen model of QC. The molecular vertex that represents perylene (Figure 1a) has the charges $Q_A = 7$ and $Q_T = 4$, which yields $Q_A - Q_T = 3$ bound states and $Q_T = 4$ resonant states. Using the model depicted in Figure 1b, we parametrized the tight-binding graph that represents the perylene terminus by fitting its reflection phase. Specifically, in the nearest-neighbor model of the linear chain, extracted and tabulated quantities include $\overline{\Omega}_0 = 3.492$ eV, $\overline{J}_1 = -0.288$ eV, on-site energy $\Omega_1 =$ 3.488 eV of the first site in the segment, and parameters of the terminus given in Table 1. The actual scattering phase, approximated by the above lattice model, can be viewed as the addition of sharp resonant features (eq 12) on top of the simplified model, in which the hopping constants J_{α} ($\alpha = 2, 3$,

6) have been set to zero due to the weak couplings between the corresponding states of the terminus and the chain (Figure 1b). The structure of the resonances of r(z) is schematically

shown in Figure 2. We obtain three poles on the real axis

 $\begin{array}{c}
1.0 \\
0.5 \\
0.6 \\
-0.5 \\
-1.0 \\
-1.0 \\
-1.0 \\
-1.0 \\
Re(z)
\end{array}$

Figure 2. Pole-zero plot of the reflection amplitude r(z) formulated in eq 6. The perylene terminus is illustrated as an example with $Q_A = 7$ and $Q_T = 3$. The poles marked as "×" are attributed to the bound states as well as the zeros ("o") in blue, whereas the zeros are related to the resonant states (in red) and unphysical solutions (in green).

corresponding to the three bound states, as well as the three zeros (in blue) associated with the poles; a "trivial" zero at z = 0 and another in the vicinity (both in green); and $Q_{\rm T} = 4$ pairs of resonances (in red) close to the circle that represent the 4 resonant states, shown as scattering phase kinks in Figure 1c. As a result, the sharp resonances have been accurately reproduced in terms of k (Figure 1c). In addition, energies of the bound states in a semi-infinite chain located outside of the exciton band can be easily found using eq 3. Although the nearest-neighbor lattice models are less accurate in terms of the exciton spectrum,²⁹ the bound state energies have been qualitatively well reproduced (Table 2).

Thus far, we have described the electronic excitations in the perylene-attached PA chain (independent of the length of the polymer) by characterizing the analytical and topological properties of the corresponding ES matrix. The aforementioned tight-binding model, which relies on the resonances between

Table 2. Bound State Energies (in eV) Relative to the Nearby Exciton Band Edge Predicted by the Nearest-Neighbor Tight-Binding Model and from QC Computations^a

bound state	1	2	3
tight-binding model	-0.312	0.028	0.143
quantum chemistry	-0.263	0.035	0.175

^{*a*}The negative value indicates the state below the exciton band, whereas positive values correspond to states above the band.



Figure 3. Electronic excitations, related to bound and resonant states, given by the contour plots of the transition density matrices from the ground state to excited states of the perylene-substituted linear PA molecule with 25 repeat units (denoted by perylene-P25), perylene, and phenylethynyl perylene (perylene-P1). The axis labels represent indices of carbon atoms starting from perylene (1–20) and along the polymer chain. The inset of each plot shows the electronic mode number, the excitation energy Ω , and the oscillator strength *f*.

the states of perylene and the exciton band of the semi-infinite PA chain, is constructed by inspecting electronic excited states in perylene-substituted PA molecules. Stated differently, the tight-binding graph representing the perylene substitution (Figure 1b) is determined by the numbers of bound and resonant excitations that we observed in the substituted linear molecules, without really performing QC analysis on the molecule of pervlene. Although the above tight-binding model excellently characterizes electronic excitations in such substituted molecules, the QC calculations found only six excitations (not seven) in perylene. Of these, only five of the six states (states 1, 2, 3, 4, and 7 in Figure 3) are considered to be relevant to the resonances in perylene-substituted PA molecules (hereafter referred to as perylene-PL, L being the length of the attached linear segment). This observation indicates that the effect of perylene substitution on the electronic excitations cannot be exactly interpreted as just resonances between states of perylene and of the chain; the analytical structure of the scattering coefficients can have substantial differences compared to the one predicted based on just resonances. Such modification of r(z) reflects the fact that perylene is "chemically" bonded rather than merely "physically" resonated with the linear segment.

Indeed, by careful examination of the transition density matrices of bound and resonant states, it is found that all of these excitations continuously extend into the first repeat unit of the attached PA segment. In particular, states 5 and 6 are highly localized on the first repeat unit of the chain rather than on the perylene (see Figure 3) and do not match any excitation in the isolated perylene. Comparing excitation structures between resonant states in perylene-PA molecules and phenylethynyl perylene (denoted by perylene-P1), we found that seven out of eight states of perylene-P1 are attributed to seven resonances in the perylene-PA molecules, whereas the remaining state of perylene-P1 is just a standing wave that resides on the triple bond of the PA, which disappears in elongated linear chains. Regarding chemical substitution, attaching perylene to a PA molecule creates a "larger room" for electronic excitations on the first PA repeat unit. Accordingly, the first repeat unit of the PA segment is more like a part of the terminus rather than of the linear chain itself in terms of electronic excitations. Regarding the tight-binding model representation, such detailed analysis on excited-state electronic structures is consequently followed by a different tight-binding morphology, that is, the terminus being described by eight lattice sites and one less site in the chain. Stated equivalently, a more "chemically" exact tight-binding model of the substituent could be built upon the substitution chemistry. It is worth mentioning that altering the structure of the lattice model, with respect to different choices of the terminus, will not affect its outcome because the relevant tight-binding parameters are adjusted by fitting the unique ES properties regardless of what is to be included in the terminus.

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In conclusion, we demonstrated that the analytical and topological structure of the reflection coefficient r(z), which can be conveniently analyzed using tight-binding (lattice) models, provides complete characterization of effects of terminal chemical substitutions in conjugated molecules on

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electronic excited-state structures. This approach can be applied to any chemical substitution in conjugated polymers, which can be treated as a molecular vertex and, thus, characterized by the ES matrix within the ES approach. By inspecting the structure of excited states, an effective tight-binding model has been formulated to incorporate the coupling between molecular substituent and the attached molecule. The tight-binding graph representing the substituent provides an analytical expression for the associated scattering matrix. By comparing the scattering matrix of the tight-binding model with the ES counterpart, we parametrize the tight-binding model and completely obtain the analytical property of the ES. Conducting the analytic continuation of the scattering amplitude to complex values of the exciton quasi-momentum, the modifications of electronic excitations are distinguished in terms of just positions of poles and zeros of the scattering amplitude that provide sufficient information on appearing bound and resonant states, respectively. Furthermore, delicate descriptions of the interaction between the chemical substituent and conjugated polymer have been attained by inspecting detailed excitedstate electronic structures. As a consequence, a corresponding tight-binding model can be built in a way not only of "phenomenological" exactness but also of "chemical" consistency.

Starting from QC data, processed by the ES method as a bridge, a chemical substitution on a conjugated polymer, in terms of electronic excitations, can be straightforwardly represented by a properly constructed tight-binding model. In return, the analytical property of ES, which adequately describes how electronic excitations in conjugated molecules are affected by the chemical substitution, can be effectively characterized. In other words, the chemical substitution effect on electronic states is fully determined by the excited-state properties of the substituent and the polymer, as well as their couplings, which have been characterized as tight-binding parameters and can provide quick and intuitive guidance for applying chemical substitutions in molecular design of organic semiconductors with desired optoelectronic properties. The described interactions between the substituent and the polymer chain play an important role in dynamical processes involving exciton-phonon couplings, for example, incoherent energy transfer and charge transport. Taking into account the simplicity of tight-binding models and their nature resting on the resonance between the substituent and the chain, the ES analysis and tight-binding representations can be feasibly applied to photoinduced dynamics in conjugated macromolecular systems.

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Notes

The authors declare no competing financial interest.

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