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Advances and Promises of Layered Halide Hybrid Perovskite Semiconductors

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ABSTRACT: Layered halide hybrid organic-inorganic perovskites (HOP) have been the subject of intense investigation before the rise of three-dimensional (3D) HOP and their impressive performance in solar cells. Recently, layered HOP have also been proposed as attractive alternatives for photostable solar cells and revisited for light-emitting devices. In this review, we combine classical solid-state physics concepts with simulation tools based on density functional theory to overview the main features of the optoelectronic properties of layered HOP. A detailed comparison between layered and 3D HOP is performed to highlight differences and similarities. In the same way as the cubic phase was established for 3D HOP, here we introduce the tetragonal phase with D_{4h} symmetry as the reference phase for 2D monolayered HOP. It allows for detailed analysis of the spin-orbit coupling effects and structural transitions with corresponding electronic band folding. We further investigate the effects of octahedral tilting on the band gap, loss of inversion symmetry and possible Rashba effect, quantum confinement, and dielectric



confinement related to the organic barrier, up to excitonic properties. Altogether, this paper aims to provide an interpretive and predictive framework for 3D and 2D layered HOP optoelectronic properties.

KEYWORDS: halide perovskites, layered materials, quantum confinement, dielectric confinement, Rashba, exciton, density functional theory, $k \cdot p$, Bethe–Salpeter equation

fter 2012, following the first convincing results in lightto-electricity conversion, $^{1-8}$ three-dimensional (3D) halide hybrid organic-inorganic perovskites (HOP) have gained a key position in solar light-harvesting applications. These materials have the general formula AMX₃, where X, M, and A are a halogen, a metallic cation (e.g., Pb^{2+}), and a "small" organic cation (e.g., CH₃NH₃⁺), respectively. Since then, HOPbased solar cells have shown growing conversion efficiencies. At the time we were concluding this review, the latest certified records exceed 21%.9-11 Importantly, excellent efficiencies have been obtained recently for cells with larger areas: 15.6 and 19.6% for a 1 cm² cell^{12,13} and 10.4% for 10.1 cm^{2,14} Combined with low-cost production and easy material synthesis, such performances place these newly born solar cells among the most promising technologies for the future. Putting aside the issue of lead toxicity, 15-17 the last obstacle on the path of HOP solar cells appears to be their instability to air and light exposure.18,19

The next evolution of HOP might come from their past with the revival of layered, that is, two-dimensional (2D), HOP structures and mixed 2D/3D structures. Indeed, due to extended chemical engineering possibilities, attractive light absorption and emission, and remarkable electronic properties at room temperature, layered HOP have increased the interest in the field of optoelectronics and microelectronics for over 30 years.²⁰⁻³⁰ Despite lower-dimensional systems having long been studied, they might also be the future of HOP-based technologies with revitalized research efforts on layered systems, including colloidal materials.^{31–37} In particular, 2D and 2D/3D Ruddlesden-Popper materials are now exhibiting improved stability and growing performances in photovoltaic devices.^{31,32,38}

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Figure 1. (a,b) Structures and (c,d) DFT band structures of $(pFPhEt-NH_3)_2PbI_4$ and $(pFPhEt-NH_3)_2SnI_4$, respectively, computed with PBE without (blue dashed line) and with (red solid line) SOC. The zero energy is referenced to the valence band maximum. Pb, Sn, I, C, F, and H atoms are depicted in gray, pink, purple, black, yellow, and white, respectively. (e) Artificially designed highly symmetric D_{4h} layered structure of $(Cs)_2PbCl_4$ and the corresponding Brillouin zone (BZ). (f) same structure doubled in the (x,y) plane and the corresponding BZ. The comparison between (g) the electronic band diagram without SOC for the initial structure and the same band diagram (h) computed using the BZ of the expanded structure highlights the band folding from the M-point to the Γ -point (see text).

Theoretical studies have closely paralleled the advancements of 3D HOP solar cells.³⁹⁻⁴² Most theoretical contributions rely on the use of density functional theory (DFT)-based simulations.^{43,44} Some of us have contributed to this effort and have shown that such atomistic calculations can be efficiently combined with complementary approaches: semiempirical models based on symmetry considerations, basic solid-state physics concepts, Landau theory of phase transitions, as well as accurate approaches to the stochastic molecular motions, which were initially developed for hybrid plastic crystals.45-49 Noteworthy, in 3D HOP, thermally activated disorder related to small organic cations is a major issue for quantitative predictions of electronic properties at room temperature using standard DFT codes.^{48,49} At low temperature, orientations of the cation are expected to be frozen and DFT calculations can be more predictive but at the price of a larger number of atoms per unit cell.³⁹⁻⁴² In contrast, the theoretical models relevant to layered HOP are much less developed. Due to very large size of the unit cells of 2D HOP structures, DFT calculations are often very limited by available computational resources. For instance, state-of-the-art DFT computations including many-body effects or DFT resolution of the Bethe-Salpeter equation⁵⁰ (BSE) for the excitonic effects are beyond our reach. Such limitations of first-principlesbased theoretical approaches are particularly detrimental, and the modeling of layered HOP remains crude and highly empirical, while their potential for technological applications is immensely broad. The dielectric mismatch between the inorganic and organic layers is expected to play a major role in optical properties and charge transport, as well as to provide a natural protection against photodegradation or moisture.³⁸ A computational modeling at the atomic scale of both layers, including the interface, is thus important.

In this review, we demonstrate that concepts and theoretical tools already well-deployed in the fields of optoelectronics and conventional semiconductors, and more recently in 3D HOP, are transferrable means to the investigation of layered HOP, as well. First and foremost, we scrutinize basic properties of the electronic structure related to reverse band ordering, as compared to conventional semiconductors, spin—orbit coupling (SOC), and quantum confinement. We then introduce a reference structure for 2D HOP, define the corresponding

Brillouin zone (BZ) folding, and illustrate related effects on real layered structures. Subsequently, structural phase transitions and related distortions of the inorganic lattice as well as their effect on the electronic band gap are examined. The possibility of symmetry inversion loss and related Rashba effect are also discussed. Finally, before concluding remarks, dielectric confinement in layered HOP is thoroughly investigated based on a recently developed nanoscale description. This introduces an alternative framework to compute excitonic effects based on the resolution of the BSE, which is currently unfeasible at the DFT level mainly because of the unit cell sizes.

Reverse Band Ordering, Spin–Orbit Coupling, and Quantum Confinement. A common feature of 3D and layered lead-based HOP is the strong SOC effect on the conduction band (CB) due to the dominant role of the metallic p orbitals of Pb.^{27,39} Notably, HOP present a reversed effect as compared to conventional semiconductors where the stronger manifestation of SOC is observed in the valence band (VB). As expected from tabulated atomic splittings, the coupling is stronger with Pb²⁺ (Figure 1c) and lesser with Sn²⁺ (Figure 1d).^{27,46} The same trend is observed in Table 1, which reports band gaps for various layered compounds including series (Bu- $NH_3_2MI_4$ (M = Pb, Sn, Ge). Surprisingly, the Ge system still presents a sizable band gap reduction of 0.15 eV with SOC. The same effect was predicted for the $Pm\overline{3}m$ phase of trichloride-based Ge 3D HOP and attributed to the SOC splitting between the p-like orbitals of total angular momentum I = 3/2 and I = 1/2, which amounts to 0.22 eV for Ge^{2+,46}.

The main difference between layered and 3D HOP is that the CB minimum (CBM) is only composed from two p-like orbitals in 2D HOP instead of three orbitals in 3D ones. Shear strain effects are inherent to layered HOP, leading to a weak degeneracy lifting and to energy minima slightly shifted away from Γ .²⁷ However, this effect is dominated by SOC, and the energy minima are retrieved exactly at the Γ -point for both split states, leading to pure direct electronic transitions (Figure 1c,d). A dramatic consequence of SOC in HOP is the density of states (DOS) reduction in the CB by a factor of 3 in 3D materials and a factor of 2 in 2D compounds.^{27,39} The DOS shrinking deeply affects both transport and optical properties. For example, optical absorptions of 3D and 2D materials are reduced by the same factors 3 and 2, respectively.²⁷ Table 1. Calculated Electronic Band Gaps E_{g} (eV), Axial (Out-of-Plane) d_{M-I}^{ax} , and Equatorial (In-Plane) d_{M-I}^{eq} Distances (Å) and Equatorial (In-Plane) Angles M–I–M (M = Pb, Sn, Ge) for (*p*FPhEt-NH₃)₂PbI₄, ^{51,52} (*p*FPhEt-NH₃)₂SnI₄, ²⁵ (Bu-NH₃)₂PbI₄, ⁵³ (Bu-NH₃)₂PbI₄, ⁵⁵ (Pentyl-NH₃)₂PbI₄, ⁵³ and (Decyl-NH₃)₂PbI₄, ⁵⁶

Organic cation	Formula	T (K)	Space group	E_g		deq	dax	MLM
				GGA	GGA+SOC	u_{M-I}	u_{M-I}	191-1-191
$F \longrightarrow NH_3^+$	(pFPhEt-NH ₃) ₂ PbI ₄	293	$P2_1/c$	2.04	1.23	3.17	3.23	153
	(pFPhEt-NH ₃) ₂ SnI ₄	293	$P2_1/c$	1.23	0.98	3.14	3.18	156
	$(Bu-NH_3)_2PbI_4$	293	Pbca	2.00	1.21	3.18	3.20	155
		223	Pbca	2.09	1.39	3.19	3.17	149
VNH ₃ ⁺	$(Bu-NH_3)_2SnI_4$	273	Pbca	1.09	0.94	3.13	3.18	159
		128	Pbca	1.33	1.15	3.17	3.14	152
	(Bu-NH ₃) ₂ GeI ₄	293	Pcmn	1.41	1.26	3.22	3.04	166
NH ₃	(Pentyl-NH ₃) ₂ PbI ₄	333	Pbca	1.95	1.16	3.19	3.20	159
		293	$P2_1/c$	2.09	1.32	3.20	3.17	154
$\bigvee_{9}^{\rm NH_3^+}$	(Decyl-NH ₃) ₂ PbI ₄	293	Pbca	1.99	1.21	3.19	3.20	156
		268	Pbca	2.17	1.37	3.17	3.19	150
		243	$P2_1/c$	2.19	1.39	3.18	3.19	150

Fortunately, since both conduction and valence bands are associated with a dispersion parabola with almost the same effective masses (*i.e.*, band curvatures), a well-balanced in-plane carrier transport is predicted for layered HOP.²⁷

In a pioneer contribution, Mitzi and co-workers proposed to consider layered HOP as type I quantum wells (QW) built from semiconducting inorganic sheets electronically confined by organic layers with much wider band gaps.²⁵ Unfortunately, this elegant representation cannot be easily transposed from the classical semiconductors to 2D HOP. Typical systems, such as para-fluorophenethylammonium-based layered HOP (pFPhEt- $NH_3)_2MI_4$ (M = Pb, Sn, Figure 1a,b), present band structures with a direct character at the Γ -point (Figure 1c,d) and no dispersion along the Γ -X direction (characterizing the stacking axis in real space), which are consistent with a type I QW. These features signify a DOS with reduced 2D dimensionality close to the band gap.²⁶⁻²⁹ However, while defining the A/B heterostructure (i.e., the layered HOP) is straightforward, the corresponding A and B bulk materials need to be delineated with caution. For instance, DFT can afford to quantitatively assess the CB and VB alignments, but the 2D HOP structure must be described as a composite material.^{28,47}

More importantly, the type I QW-like picture cannot be associated with an effective mass model to evaluate quantum confinement effects. This model predicts superlattice effects which are neither observed experimentally nor appear in DFT simulations. Another fundamental limitation of such an empirical approach comes from the computed energies of the confined charge carriers in the inorganic lattice. Those are taken from the reference 3D bulk $CH_3NH_3MX_3$ (M = Pb, Sn, and X = I, Br, Cl) but lie in a range of strong nonparabolicity. Last, in a classical semiconductor heterostructure built with various materials, effective mass model requires common Bloch basis functions. This requirement is clearly not satisfied in the case of layered HOP.²⁸

Therefore, although very deep confinement potentials are well-described and simulated, they cannot be coupled with effective mass modeling to accurately investigate quantum confinement effects. Subsequently, layered HOP must be considered as real composite materials rather than semiconductor superlattices. The discussion on quantum confinement can be carried out for HOP presenting even lower dimensions. For instance, a layered HOP with both face and corner-sharing octahedra has recently demonstrated additional confinement in one direction of the inorganic sheet due to its particular structure.⁵⁷

Brillouin Zone Folding from Reference Phases to Distorted Structures. Structural, electronic, and optoelectronic properties of 3D HOP can be understood by considering the pseudocubic $Pm\overline{3}m$ perovskite phase as a reference phase.^{39,47,48} In particular, the band structure of the Pnma and I4/mcm phases of the most popular HOP, CH₃NH₃PbI₃, are easily interpreted starting from $Pm\overline{3}m$. If the electronic band structures of the Pnma and I4/mcm phases seem very different from that of the reference $Pm\overline{3}m$ phase, plotting the band structure of the latter using the BZ of the target phases reveals the opposite. Apart from small perturbations related to an increase of the electronic band gap and a small degeneracy lifting induced by lattice strain and tilts of the inorganic octahedra, these band diagrams are directly connected to the reference's one. The electronic states close to the electronic band gap are folded from R of the cubic phase BZ to Γ of the orthorhombic supercell BZ. In addition, electronic states from M of the $Pm\overline{3}m$ BZ are also folded back to Γ .^{39,46}

Incidentally, $Pm\overline{3}m$ is experimentally observed at high temperature for the family of compounds $CH_3NH_3PbX_3$ (X = I, Br). In that case, the positions of all the atoms of the organic cations are dynamically averaged, leading to the occupation of the same spatial place at the center of the cubic cell. As far as the electronic structure is concerned, an inorganic cation, such as Cs^+ or Na^+ , can substitute the organic cation and address the disorder issue without much effect on the electronic band diagram close to the band gap.³⁹ This allows investigating the CBM and valence band maximum (VBM) electronic states, which mainly involve atomic orbitals of the inorganic octahedra.

To the best of our knowledge, no highly symmetric phase for 2D HOP has ever been reported. Therefore, the procedure applied to 3D materials cannot be directly used to investigate layered structures. An *ad hoc* attempt to tackle this issue is to design an artificial high-symmetry D_{4h} reference phase by putting Cs⁺ cations close to the inorganic layer, thus mimiking the ionic interaction (Figure 1e). The CBM and VBM are then obtained at the M-point (Figure 1g), in agreement with the early findings obtained by Knutson and co-workers on an idealized undistorted SnI₄^{2–} perovskite sheet.⁵⁸ This can be

determined from atomic orbital hybridization appearing in both VBM and CBM and showing up at the BZ edge.⁴⁶ When the M-X distance increases, the antibonding character of VBM decreases, as does the bonding character of the CBM, resulting in a larger band gap. By doubling the cell in the (x,y) plane (Figure 1f), the CBM and VBM electronic states fold back from the M- to the Γ -point (Figure 1h). Such a cell doubling is invariably observed and reveals either a symmetry reduction or octahedral tilting in the layered HOP. The actual structure, however, is established by the temperature as well as by the chemical strain of the organic part. The simplest structures are observed for some compounds, which crystallize in a monoclinic $P2_1/c$ structure (Table 1). Noteworthy, these structures contain a very large number of atoms per unit cell (over 100) exceeding by far those comprised within Pnma, R3m, Amm2, or I4/mcm structures, which are usually considered for the simulation of 3D HOP crystals. Moreover, layered HOP crystals often exhibit an additional cell doubling along the stacking axis, leading, for example, to the Pbca space group and twice more atoms in the unit cell (vide infra). Overall, these are important considerations for practical firstprinciples simulations of the layered HOP properties.

Structural Phase Transitions and Inorganic Lattice Distortions. In a structure–property perspective, one expects a direct relationship between crystal structure and band gap. Space groups and computed band gaps of several layered HOP are summarized in Table 1. It demonstrates a counterintuitive increase in band gaps for $(Bu-NH_3)_2MI_4$ (M = Pb, Sn) with decreasing temperature as a result of the reconstructive *Pbca* to *Pbca* phase transitions. Indeed, given the respective bonding and antibonding characters of CBM and VBM, a shrinking of the unit cell with decreasing temperature sing temperature should, in turn, reduce the band gap. A similar increase is predicted at the first-order reconstructive tetragonal/orthorhombic phase transition of CH₃NH₃PbI₃.^{7,8}

However, in the case of layered HOP, the specific relation between space groups involved in phase transitions is obviously not the driving mechanism of the band gap variations. For instance, the band gap of decylammonium lead iodide (Decyl- NH_3)₂PbI₄⁵⁶ undergoes a very weak decrease along the lowtemperature $P2_1/c$ phase (1.39 eV) to the intermediate Pbca phase (1.37 eV), whereas a much steeper decrease occurs at the *Pbca* to *Pbca* phase transition with $E_g = 1.21$ eV for the room temperature structure (Table 1). These trends agree nicely with those observed in the experimental photoluminescence spectra.²³ Similarly, low/high-temperature band gap variations also occur from $P2_1/c$ to *Pbca* or from *Pbca* to $P2_1/c$, depending on the length of the aliphatic chain in alkylammonium-based 2D HOP.^{23,53,56} This is due to the fact that the actual space group in 2D HOP is essentially determined by the crystal packing, the flexibility, and the thermal disorder of the organic moiety at high temperature. Thus, the resulting space group does not give direct indication on the octahedral tilting angles. In 3D HOP, the latter are known to be the origin of the band gap increase with decreasing temperature.^{46,59} Namely, due to sizable octahedral deformations, the CBM and VBM are destabilized and stabilized, respectively, thus enlarging the band gap. This behavior, well-documented for 3D materials, is yet to be observed in layered HOP. Therefore, additional temperature-dependent structural data and optical absorption spectra on 2D HOP crystals would help to rationalize the relationships between band gap and structural features.

Review

Meanwhile, a better insight in band gap evolutions of layered HOP can be gained by focusing on structural parameters of the inorganic part. Here, we report axial (*i.e.*, out-of-plane) d_{M-1}^{ax} and equatorial (*i.e.*, in-plane) d_{M-I}^{eq} distances as well as in-plane M–I–M angles for several metal iodide systems (Table 1). For instance, the slight changes in the axial and equatorial distances at the structural phase transitions in the $(Bu-NH_3)_2MI_4$ compounds with M = Pb, Sn are obviously not the critical parameters for the band gap variation. Noteworthy, all the reported in-plane M-I-M angles significantly differ from 180°, which is expected for an ideal D_{4h} reference phase (Figure 1e). This had already been discussed by Knutson and co-workers based on extended Hückel tight-binding calculations and spectroscopic data in a series $(R-NH_3)_2SnI_4$ of layered perovskites.58 They demonstrated that "the Sn-I-Sn bond angle is the dominant structural factor that controls the variation in the band gap".

To further rationalize such findings, in the same that it had been done for 3D HOP by Filip *et al.*,⁵⁹ we introduce two different angles that vanish whenever ideal D_{4h} is satisfied: (i) the in-plane projection of the M–M–X angle β (Figure 2a)



Figure 2. (a) Schematic representations of the in-plane projection of the M–M–X angle β and the δ angle chosen to characterize the in-plane and out-of-plane octahedral tilting. (b) Color map of the computed electronic band gap including SOC for a distorted (Cs)₂PbI₄ reference structure (Figure 1e) with axial d_{Pb-1}^{ax} and equatorial d_{Pb-1}^{eq} distances fixed to 3.18 Å, as a function of β and δ octahedral tilting angles. β and δ octahedral tilting angles for several layered structures are reported.

characteristic of in-plane rotation and (ii) the MX₆ octahedral tilting angle δ (Figure 2a) characteristic of out-of-plane displacement. They are defined considering an in-plane iodine atom with fractional atomic coordinate (1/4, 1/4, 0) that moves to position (1/4 – u, 1/4 + u, -w). Then, the in-plane lattice parameter a (here, we assume a = b) and angles β and δ satisfy the following trigonometrical relationships:



Figure 3. (a) Crystal structure of (PhMe-NH₃)₂PbCl₄ in the low-temperature $Cmc2_1$ phase.^{66,67} Pb, Cl, N, C, and H atoms are depicted in gray, green, blue, black, and white, respectively. (b) Band structures computed without (dashed black line) and with SOC (blue and red solid lines) in the high-temperature centrosymmetric Cmca phase (left) and low-temperature ferroelectric $Cmc2_1$ phase (right). (c) Spin textures for the inner and outer branches of the conduction band of (PhMe-NH₃)₂PbCl₄ in the $Cmc2_1$ phase. A similar behavior is computed for $(CH_3NH_3)_2Pb(SCN)_2I_2$ in the $Pmn2_1$ group. (d) Crystal structure of $(CH_3NH_3)_2Pb(SCN)_2I_2$.⁶⁸ Pb, I, S, N, C, and H atoms are depicted in gray, purple, yellow, blue, black, and white, respectively. (e) Band structures computed without (dashed black line) and with SOC (blue and red solid lines).

$$a = \frac{d_{\text{Pb-I}}}{\sqrt{(1/4 - u)^2 + (1/4 + u)^2 + (wc/a)^2}}$$
$$\tan \beta = 4u$$
$$\tan \delta = 4\frac{wc}{a}$$

where u and w are the fractional atomic displacements and c is the lattice parameter along the stacking axis. In order to highlight the influence of β and δ angles, we designed a simplified model starting from an undistorted Cs₂PbI₄ reference structure with fixed axial and equatorial distances set to $d_{Pb-I}^{ax} =$ $d_{Pb-I}^{eq} = 3.18$ Å. Figure 2b is a color map of the computed electronic band gap (including SOC) as a function of the β and δ angles. The β and δ angles for the experimental structures of (pFPhEt-NH₃)₂PbI₄ at room temperature, ^{51,52} (Bu-NH₃)₂PbI₄ at low and room temperature, $5^{3^{-}}$ and the $P2_1/c$ structure of $(Pentyl-NH_3)_2PbI_4$ and its high-temperature (333 K) Pbca structure⁵³ are reported on the color map, together with the low-temperature $P2_1/c$ structure of $(\text{Decyl-NH}_3)_2\text{PbI}_4$ as well as Pbca structures taken at 268 and 293 K.53 The corresponding band gaps (Table 1) agree nicely with the prediction of the simplified model, emphasizing the relevance of β and δ as key parameters for band gap tuning in 2D HOP. Moreover, as evidenced in Table 1 and Figure 2, the most important band gap variations between high- and low-temperature phases are related to the out-of-plane tilt angle δ . We stress that similar deformations of the octahedron are also observed along the I4/ mcm to Pnma phase transition in CH₃NH₃PbX₃ accompanied by similar band gap variations.

Loss of Inversion Symmetry and Rashba Effect. A possible splitting of the spinor bands due to a loss of inversion symmetry in 3D HOP pseudocubic structures was proposed previously by several theoretical groups.^{60–65} Various Rashba, Dresselhaus, or mixed Rashba–Dresselhaus scenarios have been introduced, which depend on the particular group–subgroup relationships between the parent centrosymmetric cubic $Pm\overline{3}m$ phase and the noncentrosymmetric child phase.⁶⁵ The spinor splittings away from the R-point of the cubic phase are predicted to be very large because the SOC effect is giant in lead-based HOP by comparison to conventional semiconduc-

tors. However, the existence of Rashba and/or Dresselhaus effect in 3D and/or 2D HOP still needs to be experimentally confirmed.

In the case of layered HOP, a similar behavior has already been predicted theoretically. (PhMe-NH₃)₂PbCl₄ crystallizes at room temperature in the ferroelectric Cmc21 phase, often encountered in layered HOP.^{66,67} This material crystallizes in a high-temperature Cmca space group (D_{2h}) , with a disordered and centrosymmetric structure. Compared to the D_{4h} reference structure, this nonpolar structure exhibits already significant distortion with an in-plane cell doubling, leading to electronic band folding from the M- to the Γ -point (Figure 1) as well as a cell doubling along the stacking axis. The $Cmc2_1$ ferroelectric phase results in a further symmetry reduction to $C_{2,w}$ compatible with a Rashba-like effect.⁶⁵ Hence, the Cmca to $Cmc2_1$ transition leads to a band splitting around the Γ -point. In a classical Rashba picture, it would be observable in the $\Gamma \rightarrow$ Y and $\Gamma \rightarrow X$ directions. However, in contrast to classical heterostructures of semiconductors, the C_2 axis is not the stacking axis. Therefore, the splitting that should be observed along $\Gamma \rightarrow X$ vanishes, while, as expected, the dispersion along $\Gamma \rightarrow Z$, that is, the high-symmetry axis, carries no splitting (Figure 3b). This specific feature leads to an unusual spin texture around Γ -point (Figure 3c) with an in-plane single direction for spins instead of the expected circulation associated with a Rashba effect.

Recently, another example of layered HOP, $(CH_3NH_3)_2Pb$ (SCN)₂I₂ (Figure 3d), has been synthesized⁶⁸ and analyzed.^{30,69,70} It belongs to the noncentrosymmetric *Pmn*2₁ group, with a S₂ axis along *c*. Hence, it can present a splitting similar to the one observed with (PhMe-NH₃)₂PbCl₄. Figure 3e displays the computed band structure.⁷⁰ The system features a slightly indirect band gap around both T (0,1/2,1/2) and R (1/ 2,1/2,1/2). Around those points, we can observe a SOCinduced splitting characteristic of a Rashba-like effect, as confirmed by the spin texture that is identical to the one obtained with (PhMe-NH₃)₂PbCl₄ (Figure 3c).

In view of this "incomplete" Rashba-like effect, one may wonder whether such an effect might ever be fully observed in a 2D layered HOP at room temperature. A first experimental demonstration of a Rashba-like effect in a layered HOP



Figure 4. (a) Slab of $(\text{Decyl-NH}_3)_2\text{PbI}_4$. Pb, I, N, and C atoms are depicted in gray, purple, blue, and black, respectively. H atoms are not displayed for clarity. High-frequency (b) dielectric $\epsilon_{\infty}(z)$ and (c) self-energy $\delta\Sigma^{2D}(z)$ profiles. (d) Optical absorption computed using the $k \cdot p/$ BSE approach for a model HOP considered as a 3D material with $\epsilon_{\infty} = 5.0$ (black line), a 2D material with only quantum confinement, and $\epsilon_{\infty} = 5.0$ (blue line), a 2D material with both quantum and dielectric confinement (red line). In the latter, the computed dielectric profile of (Decyl-NH_3)_2PbI_4 (b) is used. In all cases, the band gap is set to 2.7 eV, being the value extrapolated at room temperature from available experimental data,²³ and the reduced effective mass is set to 0.093.

structure will perhaps be realized by applying an out-of-plane external field on a nanometer scale layer of a 3D HOP as proposed by some of us recently in a spin-valve architecture.⁶⁵

Dielectric, Optical, and Excitonic Properties. The first success of layered HOP is related to the outstanding optoelectronic properties shown even at room temperature, including excellent luminescence efficiencies, white light emission,⁷¹ and nonlinear optical properties in microcavities.^{52,72} Such spectacular features are directly related to the dielectric properties of this composite material.

Following the similarity with semiconductor heterostructures, dielectric constants in 2D HOP can be analyzed with classical models relying on the abrupt interface approximation.^{73,74} However, one can go one step beyond and adopt an *ab initio* atomistic description of the material to reach the highfrequency (including only the electron gas contributions) and the static (including both the electron gas and the polar phonons contributions) dielectric profiles.^{75–78} It allows description at the atomic level and, without any empirical input, the dielectric confinement emerging from the contrast between inorganic and organic layers or the surface effects in colloidal nanostructures.⁷⁸

As a representative example of layered HOP, the highfrequency dielectric profile of $(\text{Decyl-NH}_3)_2\text{PbI}_4$ (Figure 4a) is depicted in Figure 4b. We observe clear dielectric contrast between the low dielectric constant of the organic part, ca. 2.2, and the high dielectric constant of the inorganic part, ca. 5.0. The value obtained for the organic layer is in excellent agreement with experimental data.⁷⁹ For the inorganic part, the value is slightly lower than the bulk value of the natural 3D reference material CH₃NH₃PbI₃ (*i.e.*, 5.5) but higher than the one obtained for a model slab built with one layer of the same 3D reference material (*i.e.*, 4.2).⁷⁸ This discrepancy highlights the importance of an atomic description for dielectric profiles. By including the motion of ions in the presence of an electric field, we can also compute the static dielectric constant that describes not only the effect on the electron gas but also the contribution from polar phonons. For example, in the case of $(\text{Decyl-NH}_3)_2\text{PbI}_4$, this does not affect the organic region. On

the other hand, the inorganic region magnitude increases up to 15. Meanwhile, it remains significantly smaller compared to the one obtained for the bulk 3D material CH₃NH₃PbI₃ ($\epsilon_{\infty} = 25$). This distinct behavior can be ascribed to the difference in the phonon structures for both structures.

The optical absorption is related to the dielectric constant through the Kramers–Kronig relations. In the pseudocubic phase of 3D HOP, a purely isotropic optical activity is predicted by symmetry. On the other hand, the symmetry analysis of CB and VB in layered HOP shows a pure transverse electric (TE) optical activity at the band gap energy.²⁷

A further contribution to the optical absorption close to the band gap is attributed to the presence of the excitonic electron-hole interaction. In the typical 3D HOP CH₃NH₃PbI₃, the exciton binding energy at low temperature (4 K) lies around 50 meV. However, it has been shown that this value does not describe excitonic effects at room temperature.⁴⁵ Indeed, the exciton is screened above the tetragonal to orthorhombic structural phase transition of CH₃NH₃PbI₃ (ca. 160 K).^{7,8,48} The discrepancy between low- and high-temperature experimental binding energies stems from an additional exciton screening, related to interactions between polar phonons, which leads to an increased effective dielectric constant. The thermally activated disorder of the polar cation's orientations associated with the phase transition further screens the excitonic interaction. The paradox between helium and room temperature measurements was finally resolved by the observation of the coexistence of Wannier-like and bound exciton species, observed in the 50–100 K temperature range. The helium temperature exciton line was attributed to a bound exciton with a long lifetime, possibly arising from a strong coupling with frozen orientations of the organic cations. A path to understanding of the 3D HOP excitonic structure exemplifies the need of joint spectroscopic and theoretical studies for deeper insights into optoelectronic properties of such complex materials. It is well-known that the electron-hole interaction is deeply impacted by the electron gas dimensionality. This prompts for further investigations of the excitonic properties in layered HOP.

In semiconductors or insulators, the standard approach accounting for electron—hole interaction relies on the use of the BSE built on the monoelectronic states computed at the DFT level or, even better, at the DFT+GW level.⁸⁰ Indeed, in many-body perturbation theory, the energy corresponding to a single-particle removal or addition, that is, the quasiparticle energy, ϵ^{QP} , is obtained from the GW self-energy operator in which the exchange-correlation effects are contained through the one-electron Green function, *G*, and the screening of the Coulomb interaction, *W*

$$\Sigma(\vec{r}, \vec{r}', \epsilon^{\rm QP}) = i \int \frac{dE'}{2\pi} e^{-i\delta E'} G(\vec{r}, \vec{r}', E - E') W(\vec{r}, \vec{r}', E')$$
(1)

However, in the perturbative BSE/DFT (or DFT+GW) scheme, screening of the electron-hole interaction due to atomic motion is not taken into account. Thus, BSE is more relevant at low temperature where, for example, cation motions are expected to be frozen. Moreover, from a technical point of view, the DFT+GW+BSE scheme is computationally heavy. In fact, it is roughly 100 times more numerically demanding compared to an equivalent standard DFT calculations. The DFT+GW scheme including SOC has been applied on several occasions to 3D HOP.^{60,62} More recently, Bokdam et al. used the same DFT+GW+BSE scheme with a refined description of the one-electron energies.⁸¹ Unfortunately, for complex materials such as layered HOP, which have a large number of atoms per unit cell as compared to classical semiconductors, current computational resources are generally not sufficient to afford full DFT+GW+BSE level of theory. Concerning the issue of the band gap, Fraccarollo et al. recently proposed an additive scheme where SOC and GW corrections are included separately.⁸² Here, to simulate excitonic effects in layered HOP, we suggest the use of a semiempirical framework, such as the $k \cdot p$ model combined with *ab initio* calculations.

In the literature, the enhanced exciton binding energy of layered HOP is mainly attributed to dielectric confinement effects.^{73,74} This is the case of a pioneering contribution by Muljarov *et al.*,⁷⁴ where the exciton binding energies were calculated using a classical resolution of the BSE. Unfortunately, this approach relied on an abrupt dielectric profile, and necessary parameters were fitted to experimental data obtained from bulk 3D HOP. Moreover, to avoid nonphysical divergences related to abrupt dielectric interfaces,⁷⁶ an artificial transitional layer at the interface between the organic and inorganic parts had to be considered.⁷⁴ By outlining a scheme allowing for the computation of a DFT-based dielectric profile, we herein propose an alternative strategy to model absorption properties, including excitonic effects, of 3D and layered HOP.

Starting from a reference 3D HOP, the 2D band gap can be estimated from theoretical calculations by using the following decomposition:

$$E_{\rm g} \approx E_{\rm g,DFT}^{\rm 3D} + \Sigma^{\rm 3D} + \delta E_{\rm g,DFT}^{\rm 2D} + \delta \Sigma^{\rm 2D}$$
(2)

where $E_{g,DFT}^{3D} + \Sigma^{3D}$ is the 3D band gap, including the selfenergy correction. $\delta E_{g,DFT}^{2D}$ accounts for the effects of quantum confinement and lattice distortions occurring in the 2D material. Thus, $E_{g,DFT}^{3D} + \Sigma^{3D} + \delta E_{g,DFT}^{2D}$ is the 2D band gap, in principle, available from DFT calculations. However, band gaps are usually not well-reproduced by plain DFT calculations. If more accurate strategies such as GW corrections are out of reach, one might use an experimentally determined gap or rely on the use of hybrid functionals instead. Last, $\delta \Sigma^{\rm 2D}$ contains additional corrections related to dielectric confinement.

Self-energy corrections can be computed for monoelectronic states near the band gap starting from the dielectric profile:

$$\delta\Sigma(z_0) = \frac{e_0}{4\pi} \int_0^\infty \left(V(q, z, z_0) - V_{\text{bulk}}(q, z, z_0) \right)_{z \to z_0} q \mathrm{d}q$$
(3)

where e_0 is a test charge at position z_0 and V and V_{bulk} are the electrostatic potential in the heterostructure and in the bulk, respectively. Back to GW, $\delta\Sigma(z_0)$ is equivalent to the one-particle potential profile acting on the test charge. V is then obtained by solving the inhomogeneous Poisson equation including the dielectric constant profile $\epsilon(z)$:

$$\frac{\partial}{\partial z} \left(\epsilon_{\infty}(z) \frac{\partial}{\partial z} (V(q, z, z_0)) \right) - q^2 \epsilon_{\infty}(z) V(q, z, z_0)$$

= $-e_0 \delta(z - z_0)$ (4)

A detailed description of the procedure is found in refs 76 and 78. The example of $(\text{Decyl-NH}_3)_2\text{PbI}_4$ in Figure 4c displays the self-energy profile $\delta\Sigma(z)$ computed according to the dielectric profile $\epsilon_{\infty}(z)$. At the position of the inorganic sheet, the selfenergy correction is about +0.2 eV and 10 times less than +0.02 eV for the organic part. The semiclassical evaluation of $\delta\Sigma^{2D}$, for both the CB and the VB states, can be obtained by computing the following integrals

$$\delta \Sigma_{\rm CB(VB)}^{\rm 2D} = \int \delta \Sigma^{\rm 2D}(z) \rho_{\rm CB(VB)}(z) dz$$
(5)

Here $\rho_{\rm CB(VB)}(z)$ is the electronic density profile for the CB (VB) state. The integral is computed over a unit cell, taking the central part of $\delta\Sigma^{\rm 2D}(z)$. We then obtain the total band gap correction $\delta\Sigma^{\rm 2D} = \delta\Sigma^{\rm 2D}_{\rm CB} + \delta\Sigma^{\rm 2D}_{\rm VB}$ and deduce the band gap $E_{\rm g}$ from eq 2.

Finally, to compute the full absorption spectrum including excitonic effects, we still need to include the electron-hole interaction. Here, we provide a brief description of the approach that will be described at more length elsewhere. In short, it can be implemented semiempirically by solving the BSE based on the work by Chuang *et al.*,⁸³ considering $z = z_0 =$ 0 due to the 2D character of the material of interest. The Green's function of the exciton is solved according to eq 19 of ref 83. The optical susceptibility is computed using eq 14 of the same publication. The effective mass approximation is used for the energies (eq 11 of ref 83). Those can be evaluated from the computed band structure. Corresponding wave functions, given in 8a and 8b of ref 83, are also deduced from the DFTcomputed electron densities at the center of the BZ. This scheme, in principle, calculates the electron-hole Coulombic interaction without any experimental input. In the case of typical layered HOP, evaluation of GW corrections is out of reach due to large system size. Subsequently, we can rely on the experimental data to appraise these quantities. Using again $(\text{Decyl-NH}_3)_2\text{PbI}_4$ as an example, we extrapolate the experimental band gap of 2.88 eV recorded at 1.6 K to 2.7 eV at room temperature from ref 23.

In order to demonstrate the dimensionality and dielectric effects, we use different ingredients for a 3D perovskite structure (Figure 4d, black line). The optical absorption is enhanced at the exciton resonance, and an exciton binding energy of 50 meV is computed when choosing a reduced effective mass of 0.093, a value consistent with magneto-

absorption measurements.⁸ In a second step, we restore the 2D character of $(\text{Decyl-NH}_3)_2\text{PbI}_4$ but do not take into account the dielectric properties as computed in Figure 4c. Here, the specific effect of quantum confinement on the exciton is evidenced, as shown in Figure 4d (blue line). As expected when going from a 3D to 2D structure, the exciton binding energy increases from 50 to 200 meV. When further including the dielectric confinement contribution, a large exciton binding energy of 330 meV is predicted (Figure 4d, red line), in fair agreement with available experimental data.²³ These three representative calculations illustrate the relative impacts of quantum and dielectric confinements. We emphasize that an accurate description of excitonic properties cannot avoid the precise delineation of the dielectric contrast.

CONCLUSION

The 3D HOP are current photovolatic superstars. Superb performances have motivated intense investigations and shed light on their structural, electronic, and optical properties. Meanwhile, 2D HOP, that used to be the most popular members of the family, have been in the shadow of this sudden success. However, they are coming back thanks to better photoand environmental stability in solar cells and attractive lightemission properties. Furthermore, compared to their 3D counterparts, they represent a greater playground for chemists to investigate.

In this review, we outline a systematic comparison of the structural, electronic, and optical properties of the 3D HOP and their 2D layered analogues. In both cases, SOC has a dramatic impact on the conduction bands, while it typically modifies the valence band in conventional semiconductors. A specific feature of layered materials is their quantum-well-like behavior that approaches a type I quantum confinement, however, without the superlattice effects. These composite structures are also characterized by strongly anisotropic electronic, optical, and charge carrier transport properties. Similarly to the reference pseudocubic $Pm\overline{3}m$ phase already established for 3D HOP, here, we introduce the hypothetical D_{4h} structure, which can be used to rationalize the properties of layered HOP. For instance, a comparable Brillouin zone folding can be considered to facilitate analysis of structure/property relationships. Finally, we survey dielectric and excitonic properties of layered 2D. To that end, we propose an original theoretical treatment based on an ab initio description to supersede the usual DFT+GW+BSE treatment that cannot be applied to layered HOP due to currently intractable computational cost. The former also allows for a detailed description at the atomistic level, the interface, and surface effects, however, at a greatly reduced computational

The HOP family still has an enormous potential for various optoelectronic applications, yet further progress requires intense investigations. Over the last 2 years, our understanding of 3D HOP properties has made a breakthrough thanks to the possibility of growing large monocrystals. The similar crystal engineering advance of layered HOP would also greatly benefit the field. In particular, it would allow for analysis of structures and phase transitions by enabling many spectroscopic studies (IR, Raman, Brillouin, neutron scattering, *etc.*). It would also confirm or overturn the presence of the Rashba effect that, to this moment, has not been observed in either layered or 3D HOP. The 3D structures are very constraining regarding the size of the cation, leading to severe restrictions over the choice of the organic moiety. The situation is different with 2D HOP,

where chemical engineering can follow many paths that have yet to be discovered. The wealth of HOP is not restricted to 3D and 2D systems. In this tremendous playground, many dimensions are present. For instance, among those very promising features are delivered by 3D/2D structures, where the inorganic sheet counts more than one layer of octahedra.

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Notes

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VOCABULARY

Perovskite, structure similar to the original perovskite mineral CaTiO₃ of general stoichiometry ABX₃ (respectively, A'₂BX₄) for 3D (respectively, 2D monolayered materials), where A and A' are inorganic or organic cations, B is a metallic cation, and X is an anion; Brillouin zone, primitive cell in the reciprocal space; Quantum well, 2D confined structure in which energy levels become discrete; Rashba effect, energy splitting occurring under the simultaneous presence of time-reversal symmetry and noncentrosymmetric structure, and enhanced by a strong spin-orbit coupling; Self-energy, many-body correction to a particle energy due to its interaction with the rest of the system; Exciton, neutral bound state of a system which can be related to an electron-hole pair created from the absorption of a photon; Ruddlesden-Popper perovskite, a general crystal structure of the form $A_{n-1}A'_2B_nX_{3n+1}$, where n =1 correspond to a 2D monolayered perovskite and $n = \infty$ to a 3D perovskite

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REFERENCES

(1) Kojima, A.; Teshima, K.; Shirai, T.; Miyasaka, T. Organometal Halide Perovskites as Visible-Light Sensitizers for Photovoltaic Cells. J. Am. Chem. Soc. 2009, 131, 6050–6051.

(2) Lee, M. M.; Teuscher, J.; Miyasaka, T.; Murakami, T. N.; Snaith, H. J. Efficient Hybrid Solar Cells Based on Meso-Superstructured Organometal Halide Perovskites. *Science* **2012**, *338*, 643–647.

(3) Kim, H.-S.; Lee, C.-R.; Im, J.-H.; Lee, K.-B.; Moehl, T.; Marchioro, A.; Moon, S.-J.; Humphry-Baker, R.; Yum, J.-H.; Moser, J. E.; Grätzel, M.; Park, N.-G. Lead Iodide Perovskite Sensitized All-Solid-State Submicron Thin Film Mesoscopic Solar Cell with Efficiency Exceeding 9%. *Sci. Rep.* **2012**, *2*, 591.

(4) Burschka, J.; Pellet, N.; Moon, S.-J.; Humphry-Baker, R.; Gao, P.; Nazeeruddin, M. K.; Grätzel, M. Sequential Deposition as a Route to High-Performance Perovskite-Sensitized Solar Cells. *Nature* **2013**, *499*, 316–319.

(5) Liu, M.; Johnston, M. B.; Snaith, H. J. Efficient Planar Heterojunction Perovskite Solar Cells by Vapour Deposition. *Nature* **2013**, *501*, 395–398.

(6) Nie, W.; Tsai, H.; Asadpour, R.; Blancon, J.-C.; Neukirch, A. J.; Gupta, G.; Crochet, J. J.; Chhowalla, M.; Tretiak, S.; Alam, M. A.; Wang, H.-L.; Mohite, A. D. High-Efficiency Solution-Processed Perovskite Solar Cells With Millimeter-Scale Grains. *Science* **2015**, 347, 522–525.

(7) Fang, H.-H.; Raissa, R.; Abdu-Aguye, M.; Adjokatse, S.; Blake, G. R.; Even, J.; Loi, M. A. Photophysics of Organic-Inorganic Hybrid Lead Iodide Perovskite Single Crystals. *Adv. Funct. Mater.* **2015**, *25*, 2378–2385.

(8) Miyata, A.; Mitioglu, A.; Plochocka, P.; Portugall, O.; Wang, J. T.-W.; Stranks, S. D.; Snaith, H. J.; Nicholas, R. J. Direct Measurement of the Exciton Binding Energy and Effective Masses for Charge Carriers in Organic-Inorganic Tri-Halide Perovskites. *Nat. Phys.* **2015**, *11*, 582–588.

(9) Bi, D.; Tress, W.; Dar, M. I.; Gao, P.; Luo, J.; Renevier, C.; Schenk, K.; Abate, A.; Giordano, F.; Correa Baena, J.-P.; Decoppet, J.-D.; Zakeeruddin, S. M.; Nazeeruddin, M. K.; Grätzel, M.; Hagfeldt, A. Efficient Luminescent Solar Cells Based on Tailored Mixed-Cation Perovskites. *Sci. Adv.* **2016**, *2*, e1501170.

(10) Saliba, M.; Matsui, T.; Seo, J.-Y.; Domanski, K.; Correa-Baena, J.-P.; Nazeeruddin, M. K.; Zakeeruddin, S. M.; Tress, W.; Abate, A.; Hagfeldt, A.; Grätzel, M. Cesium-Containing Triple Cation Perovskite Solar Cells: Improved Stability, Reproducibility and High Efficiency. *Energy Environ. Sci.* **2016**, *9*, 1989–1997.

(11) http://www.nrel.gov/ncpv/images/efficiency_chart.jpg (Accessed August 2016).

(12) Chen, W.; Wu, Y.; Yue, Y.; Liu, J.; Zhang, W.; Yang, X.; Chen, H.; Bi, E.; Ashraful, I.; Grätzel, M.; et al. Efficient and Stable Large-Area Perovskite Solar Cells With Inorganic Charge Extraction Layers. *Science* **2015**, *350*, 944–948.

(13) Li, X.; Bi, D.; Yi, C.; Décoppet, J.-D.; Luo, J.; Zakeeruddin, S. M.; Hagfeldt, A.; Grätzel, M. A Vacuum Flash-Assisted Solution Process for High-Efficiency Large-Area Perovskite Solar Cells. *Science* **2016**, 353, 58.

(14) Razza, S.; Di Giacomo, F.; Matteocci, F.; Cina, L.; Palma, A. L.; Casaluci, S.; Cameron, P.; D'Epifanio, A.; Licoccia, S.; Reale, A.; Brown, T. M.; Di Carlo, A. Perovskite Solar Cells and Large Area Modules (100 cm²) Based on an Air Flow-Assisted PbI₂ Blade Coating Deposition Process. *J. Power Sources* **2015**, *277*, 286–291.

(15) Babayigit, A.; Ethirajan, A.; Muller, M.; Conings, B. Toxicity of Organometal Halide Perovskite Solar Cells. *Nat. Mater.* **2016**, *15*, 247–251.

(16) Chen, P.-Y.; Qi, J.; Klug, M. T.; Dang, X.; Hammond, P. T.; Belcher, A. M. Environmentally Responsible Fabrication of Efficient Perovskite Solar Cells from Recycled Car Batteries. *Energy Environ. Sci.* **2014**, 7, 3659–3665.

(17) Chen, P.-Y.; Qi, J.; Klug, M. T.; Dang, X.; Hammond, P. T.; Belcher, A. M. Response to the Comments on "Environmentally Responsible Fabrication of Efficient Perovskite Solar Cells from Recycled Car Batteries" by Po-Yen Chen, Jifa Qi, Matthew T. Klug, Xiangnan Dang, Paula T. Hammond, and Angela M. Belcher published in Energy Environ. Sci. in 2014. *Energy Environ. Sci.* 2015, *8*, 1618– 1625.

(18) Huang, W.; Manser, J. S.; Kamat, P. V.; Ptasinska, S. Evolution of Chemical Composition, Morphology, and Photovoltaic Efficiency of CH₃NH₃PbI₃ Perovskite under Ambient Conditions. *Chem. Mater.* **2016**, *28*, 303–311.

(19) Nie, W.; Blancon, J.-C.; Neukirch, A. J.; Appavoo, K.; Tsai, H.; Chhowalla, M.; Alam, M. A.; Sfeir, M. Y.; Katan, C.; Even, J.; Tretiak, S.; Crochet, J. J.; Gupta, A.; Mohite, A. D. Light-Activated Photocurrent Degradation and Self-Healing in Perovskite Solar Cells. *Nat. Commun.* **2016**, *7*, 11574.

(20) Mitzi, D. B.; Wang, S.; Feild, C. A.; Chess, C. A.; Guloy, A. M. Conducting Layered Organic-inorganic Halides Containing < 110>-Oriented Perovskite Sheets. *Science* **1995**, *267*, 1473–1476.

(21) Kagan, C. R.; Mitzi, D. B.; Dimitrakopoulos, C. D. Organic-Inorganic Hybrid Materials as Semiconducting Channels in Thin-Film Field-Effect Transistors. *Science* **1999**, *286*, 945–947.

(22) Mitzi, D. B. Synthesis, Structure, and Properties of Organic-Inorganic Perovskites and Related Materials. *Progress in Inorganic Chemistry*; Wiley, 1999; Vol. 48, pp 1–121.

(23) Ishihara, T.; Takahashi, J.; Goto, T. Optical Properties Due to Electronic Transitions in Two-Dimensional Semiconductors $(C_nH_{2n+1}NH_3)_2PbI_4$. Phys. Rev. B: Condens. Matter Mater. Phys. 1990, 42, 11099–11107.

(24) Goto, T.; Makino, H.; Yao, T.; Chia, C. H.; Makino, T.; Segawa, Y.; Mousdis, G. A.; Papavassiliou, G. C. Localization of Triplet Excitons and Biexcitons in the Two-Dimensional Semiconductor $(CH_3C_6H_4CH_2NH_3)_2PbBr_4$. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2006**, 73, 115206.

(25) Mitzi, D. B.; Dimitrakopoulos, C. D.; Kosbar, L. L. Structurally Tailored Organic-Inorganic Perovskites: Optical Properties and Solution-Processed Channel Materials for Thin-Film Transistors. *Chem. Mater.* **2001**, *13*, 3728–3740.

(26) Sourisseau, S.; Louvain, N.; Bi, W.; Mercier, N.; Rondeau, D.; Boucher, J.-Y.; Buzaré, F.; Legein, C. Reduced Band Gap Hybrid Perovskites Resulting from Combined Hydrogen and Halogen Bonding at the Organic-Inorganic Interface. *Chem. Mater.* **2007**, *19*, 600–607.

(27) Even, J.; Pedesseau, L.; Dupertuis, M.-A.; Jancu, J.-M.; Katan, C. Electronic Model for Self-Assembled Hybrid Organic/Perovskite Semiconductors: Reverse Band Edge Electronic States Ordering and Spin-Orbit Coupling. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2012**, *86*, 205301.

(28) Even, J.; Pedesseau, L.; Katan, C. Understanding Quantum Confinement of Charge Carriers in Layered 2D Hybrid Perovskites. *ChemPhysChem* **2014**, *15*, 3733–3741.

(29) Pedesseau, L.; Jancu, J.-M.; Rolland, A.; Deleporte, E.; Katan, C.; Even, J. Electronic Properties of 2D and 3D Hybrid Organic/Inorganic Perovskites for Optoelectronic and Photovoltaic Applications. *Opt. Quantum Electron.* **2014**, *46*, 1225–1232.

(30) Saparov, B.; Mitzi, D. B. Organic-Inorganic Perovskites: Structural Versatility for Functional Materials Design. *Chem. Rev.* **2016**, *116*, 4558–4596.

(31) Smith, I. C.; Hoke, E. T.; Solis-Ibarra, D.; McGehee, M. D.; Karunadasa, H. I. A Layered Hybrid Perovskite Solar-Cell Absorber with Enhanced Moisture Stability. *Angew. Chem., Int. Ed.* **2014**, *53*, 11232–11235.

(32) Cao, D. H.; Stoumpos, C. C.; Farha, O. K.; Hupp, J. T.; Kanatzidis, M. G. 2D Homologous Perovskites as Light-Absorbing Materials for Solar Cell Applications. *J. Am. Chem. Soc.* 2015, 137, 7843–7850.

(33) Dou, L.; Wong, A. B.; Yu, Y.; Lai, M.; Kornienko, N.; Eaton, S. W.; Fu, A.; Bischak, C. G.; Ma, J.; Ding, T.; Ginsberg, N. S.; Wang, L.-W.; Alivisatos, A. P.; Yang, P. Atomically Thin Two-Dimensional Organic-Inorganic Hybrid Perovskites. *Science* 2015, 349, 1518–1521.
(34) Yaffe, O.; Chernikov, A.; Norman, Z. M.; Zhong, Y.; Velauthapillai, A.; van der Zande, A.; Owen, J. S.; Heinz, T. F.

Excitons in Ultrathin Organic-Inorganic Perovskite Crystals. *Phys. Rev.* B: Condens. Matter Mater. Phys. **2015**, 92, 045414.

(35) Tyagi, P.; Arveson, S. M.; Tisdale, W. A. Colloidal Organohalide Perovskite Nanoplatelets Exhibiting Quantum Confinement. *J. Phys. Chem. Lett.* **2015**, *6*, 1911–1916.

(36) Protesescu, L.; Yakunin, S.; Bodnarchuk, M. I.; Krieg, F.; Caputo, R.; Hendon, C. H.; Yang, R. X.; Walsh, A.; Kovalenko, M. V. Nanocrystals of Cesium Lead Halide Perovskites (CsPbX3, X = Cl, Br, and I): Novel Optoelectronic Materials Showing Bright Emission with Wide Color Gamut. *Nano Lett.* **2015**, *15*, 3692–3696.

(37) Stoumpos, C. C.; Cao, D. H.; Clark, D. J.; Young, J.; Rondinelli, J. M.; Jang, J. I.; Hupp, J. T.; Kanatzidis, M. G. Ruddlesden-Popper Hybrid Lead Iodide Perovskite 2D Homologous Semiconductors. *Chem. Mater.* **2016**, *28*, 2852–2867.

(38) Tsai, H.; Nie, W.; Blancon, J.-C.; Stoumpos, C. C.; Asadpour, R.; Harutyunyan, B.; Verduzco, R.; Crochet, J. J.; Tretiak, S.; Pedesseau, L.; Even, J.; Alam, M. A.; Gupta, A.; Lou, J.; Ajayan, P. M.; Bedzyk, M. J.; Kanatzidis, M. G.; Mohite, A. D. High-Efficiency Two-Dimensional Ruddlesden–Popper Perovskite Solar Cells. *Nature* **2016**, 536, 312–316.

(39) Even, J.; Pedesseau, L.; Jancu, J.-M.; Katan, C. Importance of Spin-Orbit Coupling in Hybrid Organic/Inorganic Perovskites for Photovoltaic Applications. J. Phys. Chem. Lett. **2013**, *4*, 2999–3005.

(40) Mosconi, E.; Amat, A.; Nazeeruddin, M. K.; Grätzel, M.; De Angelis, F. First-Principles Modeling of Mixed Halide Organometal Perovskites for Photovoltaic Applications. *J. Phys. Chem. C* **2013**, *117*, 13902–13913.

(41) Brivio, F.; Walker, A. B.; Walsh, A. Structural and Electronic Properties of Hybrid Perovskites for High-Efficiency Thin-Film Photovoltaics from First-Principles. *APL Mater.* **2013**, *1*, 042111.

(42) Giorgi, G.; Fujisawa, J.-I.; Segawa, H.; Yamashita, K. Small Photocarrier Effective Masses Featuring Ambipolar Transport in Methylammonium Lead Iodide Perovskite: A Density Functional Analysis. J. Phys. Chem. Lett. **2013**, *4*, 4213–4216.

(43) Hohenberg, P.; Kohn, W. Inhomogeneus Electron Gas. *Phys. Rev.* **1964**, *136*, B864–B871.

(44) Kohn, W.; Sham, L. J. Self-Consistent Equations Including Exchange and Correlation Effects. *Phys. Rev.* **1965**, *140*, A1133–A1138.

(45) Even, J.; Pedesseau, L.; Katan, C. Analysis of Multivalley and Multibandgap Absorption and Enhancement of Free Carriers Related to Exciton Screening in Hybrid Perovskites. *J. Phys. Chem. C* 2014, *118*, 11566–11572.

(46) Katan, C.; Pedesseau, L.; Kepenekian, M.; Rolland, A.; Even, J. Interplay of Spin-Orbit Coupling and Lattice Distortion in Metal Substituted 3D Tri-Chloride Hybrid Perovskites. *J. Mater. Chem. A* **2015**, 3, 9232–9240.

(47) Even, J.; Pedesseau, L.; Katan, C.; Kepenekian, M.; Lauret, J.-S.; Sapori, D.; Deleporte, E. Solid-State Physics Perspective on Hybrid Perovskite Semiconductors. *J. Phys. Chem. C* **2015**, *119*, 10161–10177.

(48) Even, J. Pedestrian Guide to Symmetry Properties of the Reference Cubic Structure of 3D All-Inorganic and Hybrid Perovskites. J. Phys. Chem. Lett. 2015, 6, 2238–2242.

(49) Even, J.; Carignano, M.; Katan, C. Molecular Disorder and Translation/Rotation Coupling in the Plastic Crystal Phase of Hybrid Perovskites. *Nanoscale* **2016**, *8*, 6222–6236.

(50) Salpeter, E. E.; Bethe, H. A. A Relativistic Equation for Bound-State Problems. *Phys. Rev.* **1951**, *84*, 1232–1242.

(51) Kikuchi, K.; Takeoka, Y.; Rikukawa, M.; Sanui, K. Structure and Optical Properties of Lead Iodide Based Two-Dimensional Perovskite Compounds Containing Fluorophenethylamines. *Curr. Appl. Phys.* **2004**, *4*, 599–602.

(52) Wei, Y.; Lauret, J.-S.; Galmiche, L.; Audebert, P.; Deleporte, E. Strong Exciton-Photon Coupling in Microcavities Containing New Fluorophenethylamine Based Perovskite Compounds. *Opt. Express* **2012**, *20*, 10399–10405.

(53) Billing, D. G.; Lemmerer, A. Synthesis, Characterization and Phase Transitions in the Inorganic-Organic Layered Perovskite-Type (54) Takahashi, Y.; Obara, R.; Nakagawa, K.; Nakano, M.; Tokita, J.y.; Inabe, T. Tunable Charge Transport in Soluble Organic-Inorganic Hybrid Semiconductors. *Chem. Mater.* **2007**, *19*, 6312–6316.

(55) Mitzi, D. B. Synthesis, Crystal Structure, and Optical and Thermal Properties of $(C_4H_9NH_3)_2MI_4$ (M = Ge, Sn, Pb). *Chem. Mater.* **1996**, *8*, 791–800.

(56) Lemmerer, A.; Billing, D. G. Synthesis, Characterization and Phase Transitions of the Inorganic-Organic Layered Perovskite-Type Hybrids [$(C_nH_{2n+1}NH_3)_2PbI_4$], n = 7, 8, 9 and 10. *Dalton Trans.* **2012**, 41, 1146–1157.

(57) Kamminga, M. E.; Fang, H.-H.; Filip, M. R.; Giustino, F.; Baas, J.; Blake, G. R.; Loi, M. A.; Palstra, T. T. M. Confinement Effects in Low-Dimensional Lead Iodide Perovskite Hybrids. *Chem. Mater.* **2016**, 28, 4554–4562.

(58) Knutson, J. L.; Martin, J. D.; Mitzi, D. B. Tuning the Band Gap in Hybrid Tin Iodide Perovskite Semiconductors Using Structural Templating. *Inorg. Chem.* **2005**, *44*, 4699–4705.

(59) Filip, M. R.; Eperon, G. E.; Snaith, H. J.; Giustino, F. Steric Engineering of Metal-Halide Perovskites with Tunable Optical Band Gaps. *Nat. Commun.* **2014**, *5*, 5757.

(60) Even, J.; Pedesseau, L.; Jancu, J.-M.; Katan, C. DFT and k-p Modelling of the Phase Transitions of Lead and Tin Halide Perovskites for Photovoltaic Cells. *Phys. Status Solidi RRL* 2014, *8*, 31–35.

(61) Kim, M.; Im, J.; Freeman, A. J.; Ihm, J.; Jin, H. Switchable S = 1/2 and J = 1/2 Rashba bands in Ferroelectric Halide Perovskites. *Proc. Natl. Acad. Sci. U. S. A.* **2014**, *111*, 6900–6904.

(62) Brivio, F.; Butler, K. T.; Walsh, A.; van Schilfgaarde, M. Relativistic Quasiparticle Self-Consistent Electronic Structure of Hybrid Halide Perovskite Photovoltaic Absorbers. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2014**, *89*, 155204.

(63) Amat, A.; Mosconi, E.; Ronca, E.; Quarti, C.; Umari, P.; Nazeeruddin, M. K.; Grätzel, M.; De Angelis, F. Cation-Induced Band-Gap Tuning in Organohalide Perovskites: Interplay of Spin-Orbit Coupling and Octahedra Tilting. *Nano Lett.* **2014**, *14*, 3608–3616.

(64) Stroppa, A.; Di Sante, D.; Barone, P.; Bokdam, M.; Kresse, G.; Franchini, C.; Whangbo, M.-H.; Picozzi, S. Tunable Ferroelectric Polarization and its Interplay with Spin-Orbit Coupling in Tin Iodide Perovskites. *Nat. Commun.* **2014**, *5*, 5900.

(65) Kepenekian, M.; Robles, R.; Katan, C.; Sapori, D.; Pedesseau, L.; Even, J. Rashba and Dresselhaus Effects in Hybrid Organic-Inorganic Perovskites: From Basics to Devices. *ACS Nano* **2015**, *9*, 11557– 11567.

(66) Braun, M.; Frey, W. Crystal Structure of Bis(benzylammonium) Lead Tetrachloride, $(C_7H_7NH_3)_2PbCl_4$. Z. Kristallogr. New Cryst. Struct. **1999**, 214, 331–332.

(67) Liao, W.-Q.; Zhang, Y.; Hu, C.-L.; Mao, J.-G.; Ye, H.-Y.; Li, P.-F.; Huang, S. D.; Xiong, R.-G. A Lead-Halide Perovskite Molecular Ferroelectric Semiconductor. *Nat. Commun.* **2015**, *6*, 7338.

(68) Daub, M.; Hillebrecht, H. Synthesis, Single-Crystal Structure and Characterization of $(CH_3NH_3)_2Pb(SCN)_2I_2$. Angew. Chem., Int. Ed. **2015**, 54, 11016–11017.

(69) Ganose, A. M.; Savory, C. N.; Scanlon, D. O. $(CH_3NH_3)_2Pb-(SCN)_2I_2$: A More Stable Structural Motif for Hybrid Halide Photovoltaics? J. Phys. Chem. Lett. **2015**, 6, 4594–4598.

(70) Xiao, Z.; Meng, W.; Saparov, B.; Duan, H.-S.; Wang, C.; Feng, C.; Liao, W.; Ke, W.; Zhao, D.; Wang, J.; Mitzi, D. B.; Yan, Y. Photovoltaic Properties of Two-Dimensional (CH₃NH₃)₂Pb(SCN)₂I₂ Perovskite: A Combined Experimental and Density Functional Theory Study. *J. Phys. Chem. Lett.* **2016**, *7*, 1213–1218.

(71) Dohner, E. R.; Jaffe, A.; Bradshaw, L. R.; Karunadasa, H. I. Intrinsic White-Light Emission from Layered Hybrid Perovskites. *J. Am. Chem. Soc.* **2014**, *136*, 13154–13157.

(72) Lanty, G.; Bréhier, A.; Parashkov, R.; Lauret, J.-S.; Deleporte, E. Strong Exciton-Photon Coupling at Room Temperature in Microcavities Containing Two-Dimensional Layered Perovskite Compounds. *New J. Phys.* **2008**, *10*, 065007. (73) Kumagai, M.; Takagahara, T. Excitonic and Nonlinear-Optical Properties of Dielectric Quantum-well Structures. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1989**, *40*, 12359.

(74) Muljarov, E. A.; Tikhodeev, S. G.; Gippius, N. A. Excitons in Self-Organized Semiconductor/Insulator Superlattices: PbI-Based Perovskite Compounds. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1995**, *51*, 14370–14378.

(75) Shi, N.; Ramprasad, R. Dielectric Properties of Ultrathin SiO₂ Slabs. *Appl. Phys. Lett.* **2005**, *87*, 262102.

(76) Even, J.; Pedesseau, L.; Kepenekian, M. Electronic Surface States and Dielectric Self-Energy Profiles in Colloidal Nanoscale Platelets of CdSe. *Phys. Chem. Chem. Phys.* **2014**, *16*, 25182–25190.

(77) Benchamekh, R.; Gippius, N. A.; Even, J.; Nestoklon, M. O.; Jancu, J.-M.; Ithurria, S.; Dubertret, B.; Efros, A. L.; Voisin, P. Tight-Binding Calculations of Image-Charge Effects in Colloidal Nanoscale Platelets of CdSe. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2014**, *89*, 035307.

(78) Sapori, D.; Kepenekian, M.; Pedesseau, L.; Katan, C.; Even, J. Quantum Confinement and Dielectric Profiles of Colloidal Nanoplatelets of Halide Inorganic and Hybrid Organic-Inorganic Perovskites. *Nanoscale* **2016**, *8*, 6369–6378.

(79) Richards, T. W.; Shipley, J. W. The Dielectric Constants of Typical Aliphatic and Aromatic Hydrocarbons, Cyclohexane, Cyclohexanone, and Cyclohexanol. J. Am. Chem. Soc. **1919**, *41*, 2002–2012.

(80) Hedin, L. New Method for Calculating One-Particle Greens Function with Application to Electron-Gas Problem. *Phys. Rev.* **1965**, 139, A796–A823.

(81) Bokdam, M.; Sander, T.; Stroppa, A.; Picozzi, S.; Sarma, D. D.; Franchini, C.; Kresse, G. Role of Polar Phonons in the Photo Excited State of Metal Halide Perovskites. *Sci. Rep.* **2016**, *6*, 28618.

(82) Fraccarollo, A.; Cantatore, V.; Boschetto, G.; Marchese, L.; Cossi, M. Ab Initio Modeling of 2D Layered Organohalide Lead Perovskites. J. Chem. Phys. **2016**, 144, 164701.

(83) Chuang, S.-L.; Schmitt-Rink, S.; Miller, D. A. B.; Chemla, D. S. Exciton Green's-Function Approach to Optical Absorption in a Quantum Well with an Applied Electric Field. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1991**, *43*, 1500–1509.